

Est. 1979

Harrison County Gem & Mineral Society, Inc.



Gulfport Gems

Volume 44 December 2023



Member of the Southeast Federation of Mineralogical Society

P.O. Box 10136 Gulfport, Ms. 39505



Merry Christmas & Happy 2024 New Year

www.facebook.com/gulfportgems Website: www.gulfportgems.org



A message from the President...

Dear Members,

Since we did not have a November meeting we will be voting for new officers at the Christmas Party and installing them at the January meeting. If you can't make it to the party to vote, please email your vote to the Gulfport gems address (lapidaryarts@hotmail.com).

Hope to see you all at the party. Please bring a gift to participate in the Dirty Santa drawing.

Remember, the gloves do come off when the "stealing" gets hot.

Sincerely,









Number 12

Notes from the editor . . .

Meeting:

December 16, 2023, at Diamondhead Country Club for our Christmas Lunch. See Page 2 for details.

William Holland: See Federation Page 2024 Digital Catalog is available and ready for booking. Registration opens Dec. 4, 2023. First week of class is March 31, 2024. Fed. Week 6/2/24

<u>Ballots:</u> See page 2

<u>Gems of the Month:</u>

Turquoise, Zircon, and Tanzanite

Show & Tell: None Workshops: None

2024 Workshop Instructors needed...

Please Contact Workshop Chairman, Laura at 228-342–2971

wanderingmsrn@gmail.com

Please continue to check our website and Facebook pages for updates.

Workshops are posted on both.

www.gulfportgems.org

www.facebook.com/gulfportgems







CONTRACTOR OF THE PARTY OF THE

Dec. 16, 2023

PRESIDENT NOMINEES:

ALLEN ELLIOTT 🗆 _____

VICE - PRESIDENT NOMINEES:

LISA FITCH | ______

SECRETARY NOMINEES:

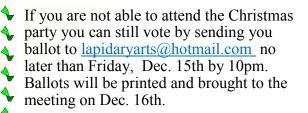
□ SHERYL ROGERS□

TREASURY NOMINEES:

□ NICOLE GREEN□

NEW DIRECTOR 1st of 3rd YR. TERM:

□ JINA MARTIN□



Board Elect for 2024

The following members will be voted on at our December meeting on December 16, 2023.

President Allan Elliott
Vice President Lisa Fitch
Secretary Sheryl Rogers
Treasurer Nicole Green

Director

Congratulations to all and "Thank You" for volunteering to serve.

Jina Martin





2024 MEMBERSHIP DUES:

Applications available at the December Christmas Party.

The deadline is the 3rd Saturday in January 2024. You may print the application from this newsletter and bring it with you.

Paying your dues on time is part of the scholarship availability



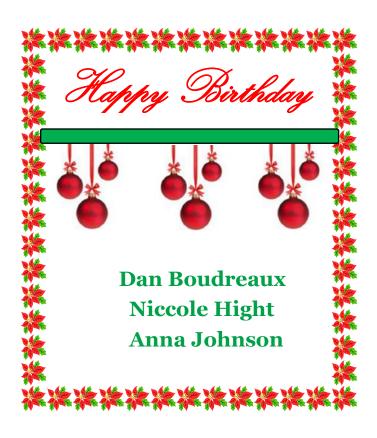
The love for all living creatures is the most noble attribute of man.

Charles Darwin



Your Editor





Off the menu . . .

ORANGE GLAZED PECANS

Ingredients:

I-I/2 cups sugar

1/4 cup Water

3 tbsp of Orange juice concentrate

2 cups Pecan Halves

1/2 tsp. grated orange zest

Directions:

In a heavy saucepan, combine the sugar, water, and OJ concentrate. Cook over medium heat until soft –ball stage. Remove from heat and stir in pecans and orange zest.

Beat until mixture thickens and loses its gloss, about 2 minutes. Drop by teaspoonfuls onto wax paper to set. Store in an airtight container.



FACTS

Mineral: Turquoise

• Chemistry: CuAl₆(PO₄)₄ (OH)_{8.5}H₂O

• Color: Blue to green

Refractive Index: 1.610 to 1.650Birefringence: Not detectable

• Specific Gravity: 2.76 (+0.14, -0.36)

Mohs Hardness: 5 to 6

11th Anniversary



Turquoise is only found in a few places on earth: dry and barren regions where acidic, copper-rich groundwater seeps downward and reacts with minerals that contain phosphorus and aluminum. The final results of this sedimentary process is a porous, semi-translucent to opaque compound of hydrated copper and aluminum phosphate.

Turquoise is believed to bring prosperity to the one who wears the stone symbolizing good fortune and success.

To chemists and geologists turquoise is as known as "copper aluminum phosphate." Often found in weathered igneous rock that contains copper minerals, where it crystallizes in veins and nodules. Developing in rocks near water tables, located in semiarid and arid environments. The chemicals in turquoise come from adjacent rock, leached out by rain and groundwater.

Being relatively soft it can be easily scratched and broken. This porous opaque stone is easily discolored by oil and pigments, and changes color when it loses some of its water content. The presence of copper gives it a sky blue color and the presence of iron gives it a greener tone.

During the formation of the stone the ochre and brown-black veins develop caused by inclusions from nearby rock fragments or from oxide staining. The most valued variety of turquoise is an intense robin's egg color or a sky blue. Hard, relatively non-porous compact stones have the best appearance because the stone can be finely polished. Pale and chalky varieties of Turquoise however are sometimes impregnated with oil, paraffin, liquid plastic and glycerin to give it a good polish.

This stone can be found in Armenia, Kazakhstan, China, Australia, Tibet, China, Mexico, Brazil, and Egypt. In Iran, where some of the best stones are found, turquoise is the national gem. The American southwest-Nevada, Arizona, Colorado, New Mexico and California-are primary producers of turquoise. Much of the specimens have a light color, and are porous and chalky-only about 10% is of gem quality.

The name originates from the French phrase "pierre turquoise" meaning "Turkish stone" because turquoise was brought to Europe by Venetian merchants who first acquired it in Turkish bazaars. When received as a gift, the turquoise symbolizes a pledge of affection because it is considered by some as a love charm. Shakespeare used this lore in "The Merchant of Venice' having Leah give a turquoise ring to Shylock when he was a bachelor. Hoping it would win his affections and persuade him to ask her to marry. Turquoise wedding rings are the popularity in Russia.

Turquoise is one of the earliest known stones to be used in jewelry even the Pharaohs of Early Egypt wore them. A tomb excavated in 1900 contained the mummified remains of Queen Zer, who ruled in 5500 B.C.; found on her arm were four magnificent turquoise bracelets. Beads dating back to 5000 B.C. have been found in Mesopotamia (now Iraq). In Iran, turquoise was the national gemstone, adorning thrones, daggers, sword hilts, horse trappings, bowls, cups, and ornamental objects.

Tanzanite

December Birthday

Anniversary—24th

FACTS



Mineral: Zoisite

Chemistry: Ca₂Al₃(SiO₄)₃(OH)
 Color: Blue to violet to bluish purple

Refractive Index: 1.691 to 1.700Birefringence: 0.008 to 0.013

Specific Gravity: 3.35Mohs Hardness: 6 to 7

Exactly who made the discovery of tanzanite is the subject of some debate. According to some, the first tanzanite crystals were found by Masai tribesman, Ali Juuyawatu. Other reports state that the discoverer was Ndugu Jumanne Ngoma. Finally, a further account credits Manual de Souza with the discovery of tanzanite on July 7, 1967. It has become the second most popular blue gem after sapphire and is rarer than diamonds.

Tanzanite is the blue to violet to purple variety of the mineral zoisite. It is mined commercially only in one area of the world: the Merelani Hills of Tanzania, which is where it got its name by Tiffany & Co. Because crystals show different colors depending on viewing direction, cutters can choose bluish purple or the more favored pure blue or violetish blue hue depending on how much weight they want to retain from the rough.

The blue color of tanzanite is caused by small amounts of vanadium within the zoisite mineral structure. When vanadium-bearing zoisite is heated to a temperature of 600 degrees Celsius for about 30 minutes, the oxidation state of the vanadium is changed and that change causes or improves the blue color. Tanzanite is a pleochroic gem.

Pleochroism is a physical property in which the material appears to be different colors when viewed from different crystallographic directions. Some specimens of tanzanite can be a distinct blue when viewed from one direction, and vary from violet to red when viewed from other directions.

Gircon

<u>FACTS</u>

Mineral: ZirconChemistry: ZrSiO₄

• Color: Blue, red, yellow, orange, brown, green

• Refractive index:

High: 1.925 to 1.984 (+/- 0.040)
Medium: 1.875 to 1.905 (+/- 0.030)
Low: 1.810 to 1.815 (+/-0.030)

• Birefringence: 0.000 to 0.059 (low to high)

• Specific gravity: 3.90 to 4.73

• Mohs Hardness: 6 to 7.5 (low to high)

Colorless zircon is known for its brilliance and flashes of multicolored light, called fire. These zircon properties are close enough to the properties of diamond to account for centuries of confusion between the two gems. Colorless zircon is called "Matara" zircon after a city in Sri Lanka near where it is mined.

Zircon occurs in an array of colors. Its varied palette of yellow, green, red, reddish brown, and blue hues makes it a favorite among collectors as well as informed consumers.

The crystals grow in many different types of rock and possess a range of optical and physical properties.

Some lower value zircons—usually green ones—were determined by scientists that the crystal structures of these gems were almost completely broken down by radioactive elements—often present in zircon as impurities—that damaged the gems' crystal structure over long periods of geological time.

Gemologists classify zircons into three types—high, intermediate, and low depending on the its properties, which are directly related to the amount of radiation-induced damage done to its crystal structure.

High or normal zircons have full crystal structures, with little or no damage from radioactive elements possessing the normal physical and optical properties associated with the mineral.

In intermediate or medium zircons, radioactive elements have caused some structural damage possessing physical and optical properties that are between high and low types.

All the zircons used in jewelry are of the high type. Interestingly, radiation-induced crystal-structure breakdown can be reversed somewhat by heating zircon to high temperatures this repairs the stone's damaged crystal structure producing a high type gemstone.



Sources: gia.edu, Wikipedia, geology.com, gemselect.com, minerals.com, durangosilver.com, smithsonianmag.com, google.com







TESTING FOR SILVER

Often you need to identify some of those unknown "silvery" pieces of metal in the bottom of the toolbox or some piece of old jewelry that is not hallmarked. Is it silver or is it something else?

Of course, if you need to know exactly what you have, it's best to send your metals off for refining. But inexpensive silver testing solutions can be used to help distinguish higher silver content alloys from alloys that have the same appearance but with little to no silver content, like German Silver or Nickel.

I purchased a half-ounce bottle of JSP Silver Testing Solution #GT41. It's not a rigorous analytic test, but it lets you know if you're on the right track. And it's inexpensive. Mine was only \$3.

With a fresh solution you have an instant reaction after applying it to the metal being tested. The procedure is simple - apply a small drop and watch for a color change. Note that the acid will leave a slight mark, so choose a spot that is out of the way or will be easy to polish.

If you suspect the object is silver plated, you should file a little notch somewhere inconspicuous to expose what metal is below the surface. Otherwise, all you test will be the surface plating.

Here's the reaction I got when testing various materials:

. Fine silver	. Red/Orange	. Sterling silver	Brick Red
. 80% silver 20% copper	r . Dark red changing to gray	. Brass	. Yellow changing to blue
. Nickel	. Gray-green	. Copper	. Yellow changing to blue
. Steel	. Black	. Stainless Steel	No color change

Caution - If you do any of this testing, know that you are handling a reasonably strong acid. The GT41 label says it includes nitric acid and potassium dichromate.

- * Wear safety glasses.
- * Use a solution of baking soda and water to neutralize acid.
- * Do not get any testing solution on your skin.
- * Wash and clean up well when you're done.

Making jewelry involves a multitude of skills, intricate hand work, and a lot of problem solving. In this series Books in this series help to::

- >Broaden your metalworking skills
- >Improve productivity at the bench
- >Save money on tools and supplies

You'll find hundreds of low cost and really practical tips and techniques that the author uses in his work and teaches in his classes and workshops.

http://amazon.com/dp/B0BQ8YVLTJ





Smart Solutions for Your Jewelry Making Problems

http://amazon.com/author/bradfordsmith

www. bradsmithjewelry.com
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Southeast Federation News

American Federation News



S.F.M.S. Newsletter

Subscription to Lodestar is FREE electronically

Email: sfms.lodestar@gmail.com Or

Editor: Lori Heinemann

Can be read online:

http://www.amfed.org/sfms/index.html

www.amfed.org

SFMS 2023 Annual Convention

March 11, 2023 Live Oak, FL

William Holland School of Lapidary Arts 706-379-2126

Register Now at:

www.lapidaryschoolregistration.org

Federation Week Jun 2 - 8, 2024

Classes are subject to change

WILDACRES RETREAT 828-756-4573

Website @www.wildscres.org

registrarwildacres@gmail.com Claudia Erwin

When scheduling keep in mind, workshops are subjected to change without notice.

Please check the website for changes.

Federation Weeks August 4th - 20th

2nd Week September 11th -17th

Nearby Gem Societies

Ms. Gulf Coast Gem & Mineral Society www.mgcgms.com

Meeting: 2nd Saturday of the month

Mobile Rock and Gem Society
www.mobilerockandgem.com
Meeting: 2nd Tuesday of the month

Gem & Mineral Society of Louisiana Meeting: 2nd Monday of the month

Contact: Levette @ 504-214-3205

A.F.M.S. Newsletter

Subscriptions are \$4.50 per year
Checks should be made payable to "AFMS"
AFMS Central Office

Steve Weinberger PO Box 302 Glyndon, MD 21071-0302

central office@amfed.org PH: 410-833-7926

AFMS 2023 Annual Convention

October 28-29, 2023

St. Lucie County Gem Show in Port St Lucie, FL

A.L.A.A.

Have you joined the **American Lands Access Association** (ALAA)? Help is needed. Please volunteer!

The organization was founded to promote and ensure the right of the amateur hobby collecting, recreational prospecting and mining, and the use of public and private lands for educational and recreational purposes and to carry the voice of all amateur collectors and hobbyists to our elected officials, government regulators and public land managers.

Individual dues are \$25 per year; clubs \$50 per year.

A quarterly newsletter is sent to all members. For more information,

Visit http://amlands.org

Shirley Leeson, President



Other Interests:

International Society of Glass Blowers
Visit site if you like Lampworking

Website: www.isgb.org



Harrison County Gem & Mineral Society Information



Officers 2023

<u>President</u> - Allan Elliott Allan.K.Elliott@gmail.com <u>Vice President</u> - Lisa Fitch lapidaryarts@hotmail.com

Secretary - Rosalind Norvel-Daniels

Dzignsbyrozz@gmail.com

<u>Treasurer</u> - Nicole Green nfouasnon@gmail.com

Directors

Lynn Tate (3 yr.) l.tate4@gmail.com Sharon Greer (2 yr.) 3catsgreer@bellsouth.net Reba Shotts (1 yr.) Buddynrebashotts@bellsouth.net

Committee Members

Workshop & Facebook Chairman

Laura Tate wanderingmsrn@gmail.com

<u>Equipment Chairmen</u>

Cindy Braden 228-243-2665

Field Trip Chairman

Sharon Greer 3catsgreer@bellsouth.net

Librarian Chairman

Sue West msuewest@bellsouth.net

Newsletter Editor

Lisa Fitch lapidaryarts@hotmail.com

Sunshine & Hospitality Chairman

Clemencia Howe clemenhowe@hotmail.com

Show Chairmen

Lynn Tate 1.tate4@gmail.com

Webmaster & Workshop Chairman

Leslie Jancovich ljancovich@gmail.com

Assistant Webmaster

Nicole Green nfouasnon@gmail.com

Southeast Federation of Mineralogical Societies

Ms. State Director, Parliamentarian & Boundaries

Leslie Lane 601-344-8171

Past President SFMS

John Wright osjbw@cableone.com

Meetings / Workshops

Herbert Wilson Recreational Center 3625 Hancock Avenue Gulfport, Ms. 39501

3rd Saturday of each month

9 am - 12 pm Workshop 12 pm - 12:30pm Program 12 pm - 1 pm Lunch 1 pm - 2 pm Meeting

<u>Directions:</u> Coming from Pass Road or Hwy. 90 turn on Courthouse Road.

Turn next to Hancock Bank on 33rd Street until you reach Hancock Street. Look to your left and you will see the Herbert Wilson Recreational Center. We are on the left side of building thru the gates.

Come for fun . . . Stay to play!

Visit our Website and Facebook page:

www.gulfportgems.org or https://www.facebook.com/gulfportgems

2023 MEETING DATES

Jan 21st May SHOW 20th & 21st

Feb 18th June 17th Sept 16th Mar 18th July 15th Oct 21st April 15th Aug 19th Nov 18th

Dec 16th Christmas Party

Newsletter Editor



Lisa Fitch

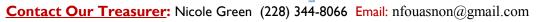
Submit newsletter articles to: lapidaryarts@hotmail.com

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Articles and minutes are due by the last day of each month



Membership Form





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Our goal is to promote, educate and stimulate interest in the earth sciences and lapidary arts context.



Address Service Requested

Merry Christmas





THE OFFICIAL PUBLICATION FOR THE HARRISON COUNTY GEM & MINERAL SOCIETY, INC.

